



PAMIBIA UNIVERSITY
OF SCIENCE AND TECHNOLOGY

FACULTY OF HEALTH AND APPLIED SCIENCES

DEPARTMENT OF NATURAL AND APPLIED SCIENCES

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| QUALIFICATION: BACHELOR OF SCIENCE (MAJOR AND MINOR) | |
| QUALIFICATION CODE: 07BOSC | LEVEL: 7 |
| COURSE CODE: BPH702S | COURSE NAME: BIOMEDICAL PHYSICS |
| SESSION: JANUARY 2019 | PAPER: THEORY |
| DURATION: 3 HOURS | MARKS: 100 |

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| SUPPLEMENTARY/SECOND OPPORTUNITY EXAMINATION PAPER | |
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| MODERATOR: | MRS. NDESHIHAFELA VERA UUSHONA |

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| INSTRUCTIONS | |
| 1. | Write all your answers in the answer booklet provided. |
| 2. | Read the whole question before answering. |
| 3. | Begin each question on a new page. |
| 4. | Useful constants are listed on the last page for this question paper. |

PERMISSIBLE MATERIALS

Scientific Calculator

ATTACHMENTS

None

THIS EXAMINATION PAPER CONSISTS OF 5 PAGES

(INCLUDING THIS FRONT PAGE)

QUESTION 1**[20]**

- 1.1 Explain the following terms: (i) oncology (ii) computed tomography. (4)
- 1.2 Differentiate between ultrasound and magnetic resonance imaging modalities. (6)
- 1.3 The compressive strength of a bone is $1.8 \times 10^{10} \text{ N/m}^2$. Estimate the compressive strain of the bone which is 15 cm long and is compressed by 1.6 mm. Also calculate the elasticity of the bone after 1.6 mm compression. (6)
- 1.4 Define diffusion and state Graham's law of diffusion. (4)

QUESTION 2**[20]**

- 2.1 Estimate the specific ionization resulting from the passage of a 0.25-MeV beta particle through standard air, given that the mean ionization for air is 34 eV/ip. (10)
- 2.2 From (2.1), what is the relative (to air) mass stopping power of graphite, density = 2.26 g/cm³, for a 0.25 MeV beta particle.
- The density of standard air is $1.293 \times 10^{-3} \text{ g/cm}^3$. (4)
- 2.3 A nozzle with a radius of 0.250 cm is attached to a garden hose with a radius of 0.900 cm. The flow rate through hose and nozzle is 0.500 L/s.
- a) Calculate the speed of the water in the hose, (4)
- b) Write down Bernoulli's equation for an incompressible, frictionless fluid. (2)

QUESTION 3**[20]**

- 3.1 Given that the ultrasound frequency of 2 MHz and wavelengths of 6.5 nm was used to take an image of a bone. Calculate the acoustic impedance of a bone if the density of a bone is $2.2 \times 10^3 \text{ kg/m}^3$. (4)

3.2 If the intensity of reflected ultrasound is $R = \left(\frac{Z_2 - Z_1}{Z_2 + Z_1}\right)^2$, show that the intensity for transmitted ultrasound wave is given by $T = \frac{4Z_1 Z_2}{[Z_1 + Z_2]^2}$. (6)

3.3 The relative sound intensity is measured on a logarithmic scale. Calculate the relative intensity of a sound beam is;

(i) reduced by 45%. (5)

(ii) a two-fold increase. (5)

QUESTION 4 [20]

4.1 State and discuss the law of radioactive decay. (5)

4.2 Sodium (Na) has a half-life of 15 hours. Calculate the activity of a 30-MBq source of Na-24 after 2.5 days. What is its decay constant? (5)

4.3 Explain how do you understand the following terms: (4)

(i) Dose rate (D)

(ii) Equivalent dose (D_e)

4.4 A Mo99m/Tc99m Generator is in transient equilibrium. The radioactivity of Mo-99m at time t_0 is 16 GBq. After 156 hrs the activity of mo-99m is 3.2 GBq if no milking takes place. Estimate the activity of the daughter nuclide. Note: Mo-99m $T_{1/2} = 67$ hours and Tc-99m $T_{1/2} = 6$ hours. (6)

QUESTION 5 [20]

5.1 During radiotherapy some non-radioactive sources may be used for efficient treatment of cancer, like **x-ray tube devices** and **Linear Accelerators**. Discuss the structures of these sources. (6)

5.2 In Windhoek Central Hospital a high-dose-rate remote after-loading brachytherapy machine is used. Explain how the after-loader machine operates. (4)

Some useful constants/formulae:

- Speed of light $c = 3.0 \times 10^8$ m/s
- The linear rate of energy loss due to excitation and ionization is ;
- $$\frac{dE}{dx} = \frac{2\pi q^4 NZ (3 \times 10^9)^4}{E_m \beta^2 (1.6 \times 10^{-6})^2} \left\{ \ln \left[\frac{E_m E_k \beta^2}{I^2 (1 - \beta^2)} \right] - \beta^2 \right\} \frac{MeV}{cm}$$
 and $E_k = m_o c^2 \left[\frac{1}{\sqrt{(1 - \beta^2)}} - 1 \right]$,
where
- q is the charge on the electron, $1.6 \times 10^{-19} C$,
- N is the number of absorber atoms per cm^3 ,
- Z is the atomic number,
- NZ is of absorber electrons per $cm^3 = 3.88 \times 10^{20}$ for air at 0° and 76 cm Hg,
- E_m is the energy equivalent of the electron mass, 0.51 MeV,
- E_k is the kinetic energy of the beta particle in MeV,
- β is the speed of the ionization particle/speed of light, $\beta = v/c$,
- I is the mean ionization and excitation potential of absorbing atoms (MeV), $I = 8.6 \times 10^{-5}$ for air and $I = 1.35 \times 10^{-5} Z$ for the substance.